



S. N. Patel Institute of Technology & Research Centre, Umrakh

(A Vidyabharti Trust Institution)

BE/Chemical Engineering

Subject Name: Chemical Reaction Engineering-I

Subject Code: 3160506

Sr. No.	Experiment
1	To determine the activation energy of the reaction between sodium thio-sulphate and HCl using Arrhenius Equation.
2	To measure the kinetics of a reaction between ethyl acetate and sodium hydroxide at room temperature by integral method of analysis.
3	To measure the kinetics of a reaction between ethyl acetate and sodium hydroxide at room temperature by differential method of analysis.
4	To measure the kinetics of a reaction between ethyl acetate and sodium hydroxide at room temperature by half-life method of analysis.
5	To determine the reaction rate conversion for the given saponification reaction of ethyl acetate and NaOH solution in the isothermal batch reactor
6	To study and analyze Residence Time Distribution (RTD) for three tanks in series.
7	To determine the conversion in PFR for saponification of ethyl acetate with NaOH at ambient condition.
8	RTD studies in a CSTR.
9	To study of a non-catalytic homogenous reaction in a coil type plug flow reactor under isothermal condition.



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BE/Chemical Engineering

Subject Name: Mechanical Operation

Subject Code: 3150502

Sr. No.	Experiment
1	Jaw Crusher
2	Roll Mill
3	Sieve Shaker
4	Cyclone Separator
5	Plate and Frame Filter
6	Froth Flotation
7	Batch Sedimentation
8	Basket Centrifuge